

Improving Adhesive Bonding of Composites

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- Industry Participation
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Two projects will presented today

1. Amine Blush in Epoxy Paste Adhesives
 - Characterization of Bondline



Amine Blush in Epoxy Paste Adhesives

- Project Background
- Test Methods and Design
- Current Results
- Upcoming Work



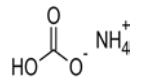
Motivation and Key Issues

- Bond failures have been attributed to amine blush
 - Cessna incident (2010)



- What are the conditions for blush formation?
- What are the effects on bond quality?
- Investigate blush NDI methods





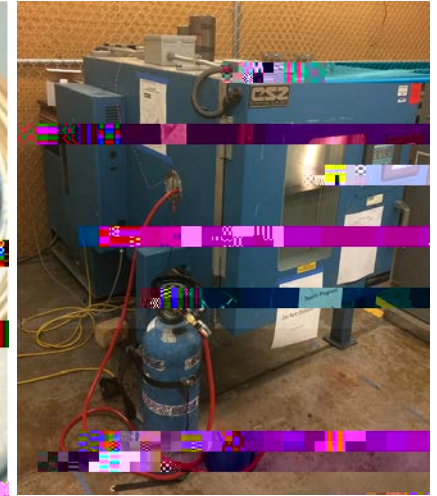
Project Outline

1. Expose epoxy paste bonds to a range of environmental conditions and analyze the effect on bond strength.
2. Perform FTIR analysis on traveler coupons to determine if blush formation can be detected and quantified.
3. Relate the mechanical response of the bond relative to any measured blush formation.



Project Test Plan

1. Adhesive: Henkel Loctite EA 9360
 - Two part epoxy – amine system
2. Experimental Exposure Variables
 - Temperature (60 – 90 F)
 - Humidity (50 – 85 %RH)
 - CO₂ (500 – 5000 ppm)
 - A & B component exposure time (0 – 3 weeks)
 - Bondline exposure time (20 – 50 min)
 - 2⁵⁻¹ Factorial DOE
3. Test Methods
 - FTIR – Amine Blush Measurement
 - T-Peel – Bond Quality Measurement
 - DCB – Bond Quality (In progress)



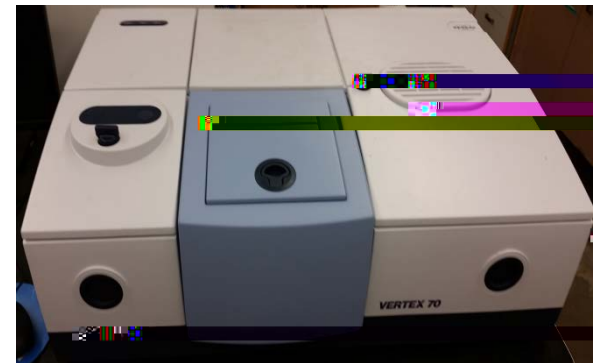
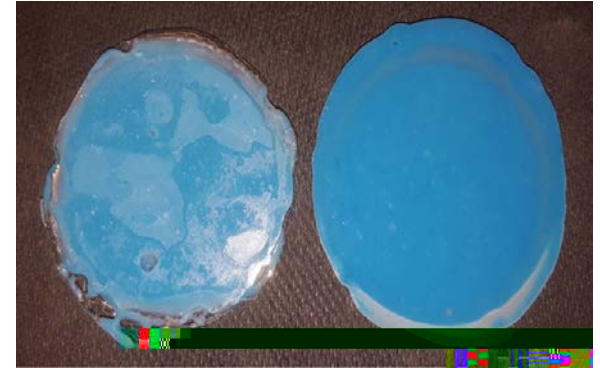
Exposed Amine Component (Left)
Environmental Conditioning Chamber (Right)





FTIR Analysis

- Testing with benchtop ATR FTIR
 - Samples exposed alongside mechanical samples
 - Samples closed out after designated exposure time
- Establishment of quantification procedures
- Comparisons to reference literature

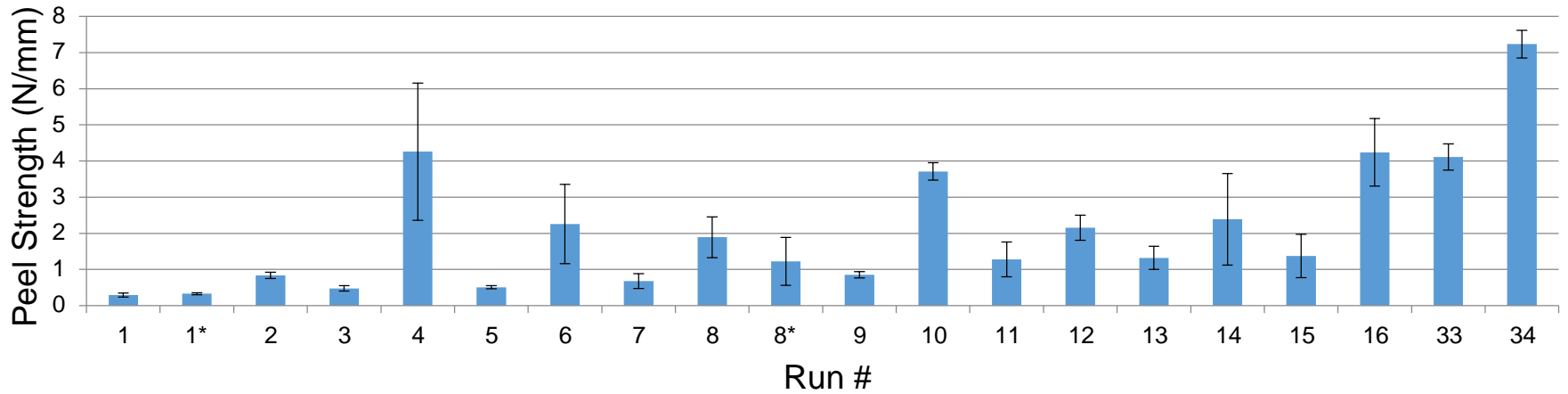
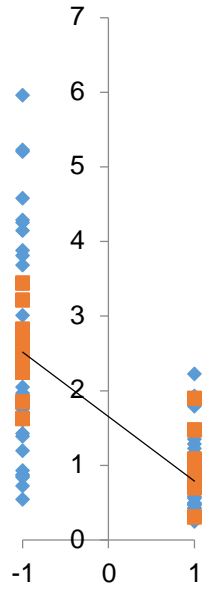


T-



Mechanical Testing Results

T-Peel e B 92.136 334.68 5.04 -5.04 re B 27



Mechanical Testing Results

1. Temperature was strongest factor for decrease in strength
2. Other significant negative factors:
 - Bondline Time Exposure
 - Part A&B Time Exposure
3. CO₂ had no measureable effect on t-peel strength
4. Humidity (50-85%) had no measurable effect on t-peel strength.
5. Combination effects are being analyzed.

Regression Statistics

Multiple R	0.759
R Square	0.576
Adjusted R Square	0.543



FTIR Results



- Notable difference between control and exposed samples



FTIR Results



Discussion

- How can we correlate FTIR measurements to mechanical properties?
 - Can FTIR Traveler coupons represent bondline?
 - f* Direct measurement of fracture surface.
 - f* Measure FTIR before and after cure.
 - Blush Thickness
 - f* FTIR detection may not reflect the amount (thickness) of blush.
 - f* May be detecting thin layers that are not detrimental to bond quality.
 - Bonding Procedure
 - f* Adhesive squeeze out may break up thin layers of blush



Looking forward- Amine Blush



MOTIVATION AND KEY ISSUES

- Most important step for bonding is surface preparation



MEASURING SURFACE ENERGY



IGC VS. CONTACT ANGLE

Contact Angle (CA)

- Small drops (1 ml) of 3-5 known liquids placed on surface
- Surface energy calculated over small area (order of mm²)
- Can be affected by surface texture (non-circular drops)
- Quick, inexpensive, can be portable

Inverse Gas Chromatography (IGC)

- 8-10 Known gases flow over surface
- Larger area sampled (2"X8")
- More information obtained (higher fidelity data)
- Distribution of surface energy
- Greater sensitivity to subtle changes
- Expensive equipment, skilled operator



IGC METHODOLOGY

Variable	Description
Probe Gases	Undecane, Decane, Nonane, Heptane, Dichloromethane, Ethyl Acetate, Acetonitrile, Acetone
Targeted Fractional Surface Coverage	

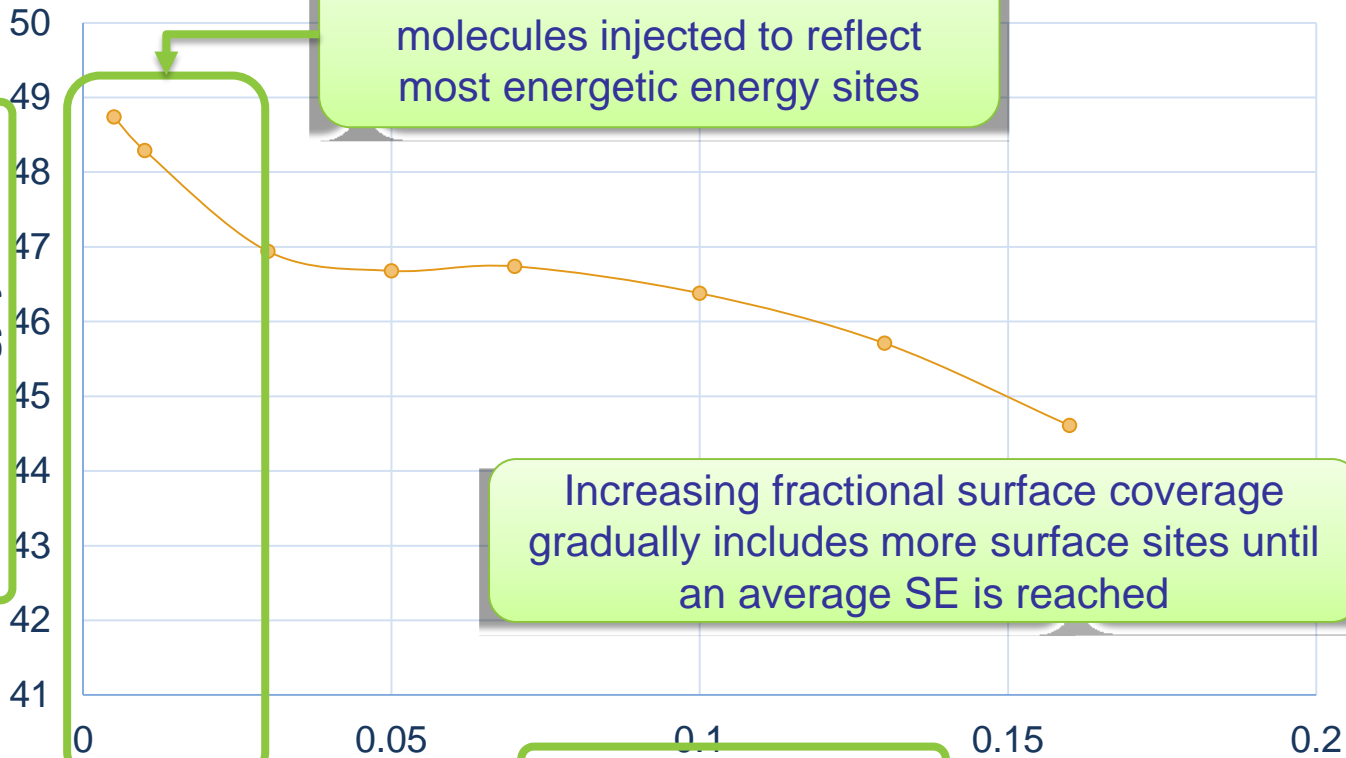
0.01,



IGC Surface Energy Profiles

Total Surface Energy

Surface Energy, γ mJ/m²



Infinite dilution – small number of molecules injected to reflect most energetic energy sites

Increasing fractional surface coverage gradually includes more surface sites until an average SE is reached

Coverage, n/n_m

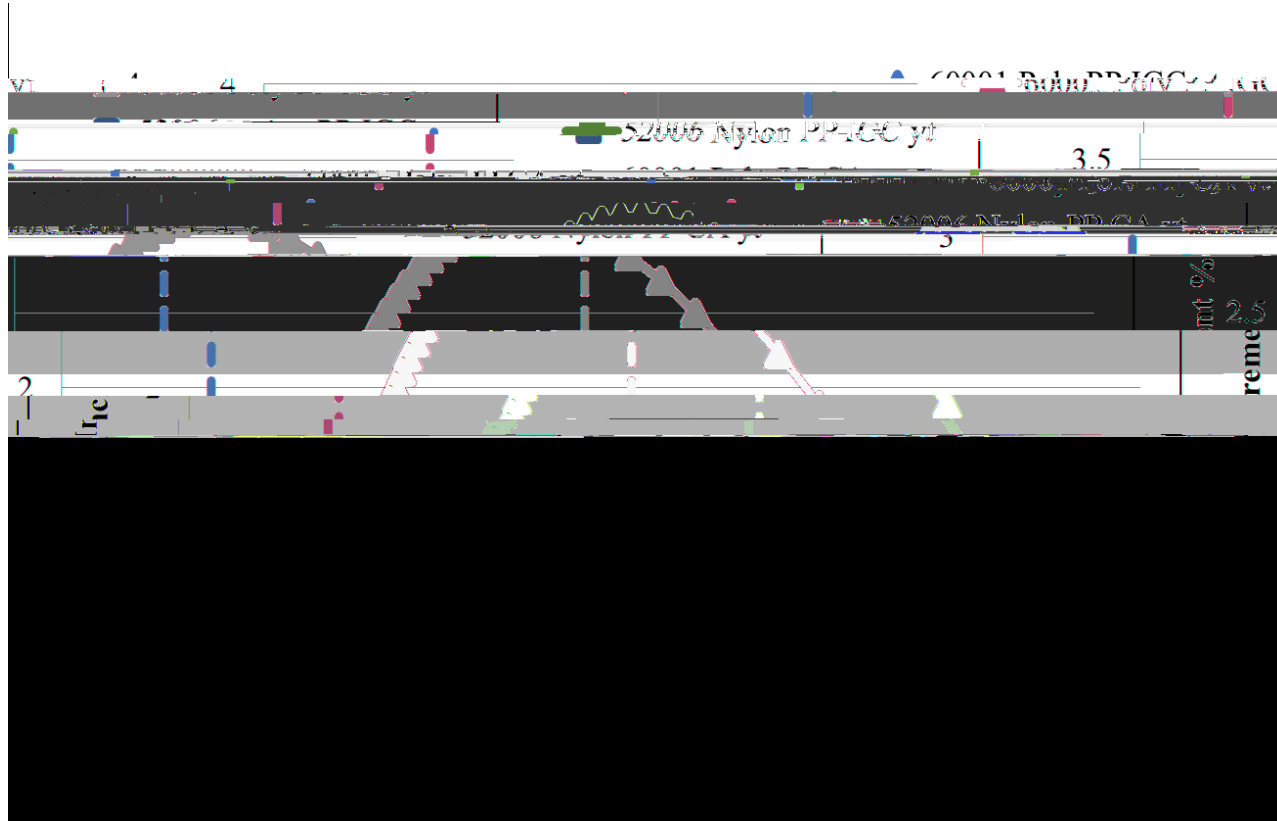
Theoretical fractional surface coverage of the monolayer with ratio of injected moles to moles required to cover the surface

#2 T800/3900 & 60001 2Hr



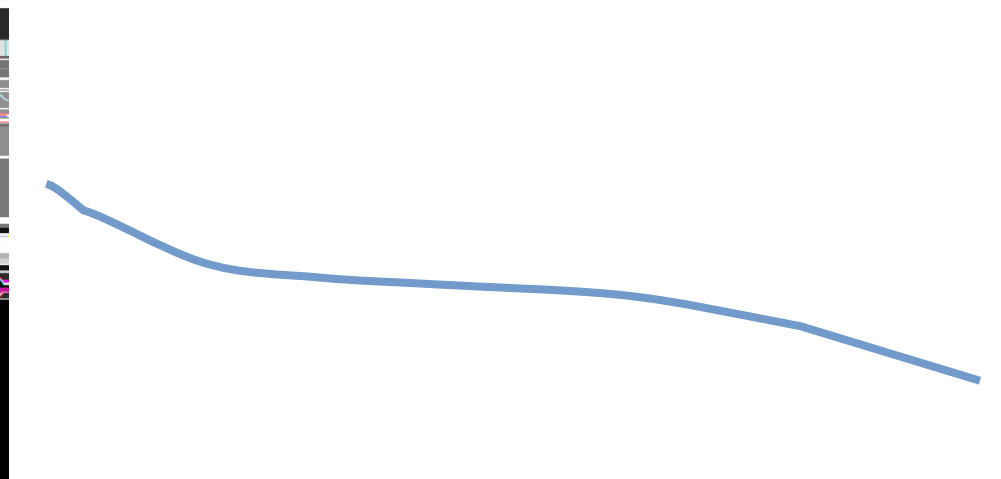


IGC AND CA COMPARISON



1. Nylon and Polyester have significantly different distributions according to IGC
2. Contact angle is controlled by complex wetting phenomena
3. Contact angle correlation to the IGC data is different for each peel ply type

ICCP 121 - Sustainable Development



Pilot Study Conclusions/Discussion

- Contact angle measurements result in a single value of surface energy (

Current Work

Priority
Testing

- Understand the advance models of wetting versus gas interactions
- Characterize additional surface preparation methods with IGC
- Relate surface preparation to bond quality types
- Additional statistical data and material coupon testing for a more complete representation of the bonding surface
 - X-ray photoelectron spectroscopy (XPS)
 - Scanning electron microscopy (SEM)
 - Double cantilever beam (DCB)

Although IGC is able to provide more information on surface energies related to various surface preparations techniques, other components contributing to the quality of the bonding surface need to be investigated.

Current Work

- Surface Preparation Methods (in order of research priority)
 - Plasma Treated
 - Grit Blasted
 - Orbital Sanding
- Surface Energy Testing
 - IGC
 - Advanced Contact Angle Measurements
 - Static, advancing and receding

2016	2017		
Q4	Q1	Q2	Q3
Materials Selected	Panel Fabrication		

Current Work

Panel #	Substrate	Surface Prep	Cure Dwell	Comments
1	Toray's 3900/T800 Unidirectional Prepreg	60001	2	Baseline for plasma treatment
2	Toray's 3900/T800 Fabric Prepreg	60001	2	Baseline for plasma treatment
3, 4, 5	Toray's 3900/T800 Unidirectional Prepreg	60001 & Plasma Treated	2	3 panels for 3 separate IGC runs
6, 7, 8	Toray's 3900/T800 Fabric Prepreg	60001 & Plasma Treated	2	3 panels for 3 separate IGC runs
9	Toray's 3900/T800 Unidirectional Prepreg	FEP	2	Baseline for sanding and grit blasting
10	Toray's 3900/T800 Fabric Prepreg	FEP	2	Baseline for sanding and grit blasting
11, 12, 13	Toray's 3900/T800 Unidirectional Prepreg	FEP & Grit Blasted	2	3 panels for 3 separate IGC runs
14, 15, 16	Toray's 3900/T800 Fabric Prepreg	FEP & Grit Blasted	2	3 panels for 3 separate IGC runs
17, 18, 19	Toray's 3900/T800 Unidirectional Prepreg	FEP & Orbital Sanded	2	3 panels for 3 separate IGC runs
20, 21, 22	Toray's 3900/T800 Fabric Prepreg	FEP & Orbital Sanded	2	3 panels for 3 separate IGC runs
23, 24, 25	Toray's 3900/T800 Fabric Prepreg	60001 & Plasma Treated	6	Contingency testing if time/funding permits

Priority Testing



Looking Forward

Surface Characterization using iGC Methods

- Benefit to Aviation
 - Better understanding of surface prep.
 - Guide development of QA methods for surface prep.
 - Greater confidence in adhesive bonds
- Future needs
 - Surface energy (wetting) vs. bond quality
 - Surface energy at cure temperature
 - QA method to ensure proper surface for bonding
 - Applicability to other composite and adhesive systems
 - Model to guide bonding based on characterization, surface prep. and material properties





Questions?



DOE Test Matrix

Control	Low Level (-1)	High Level (+1)	Factor



Potential Areas to Explore

1. Bonding procedure
 - Mound vs even spread
 - Using vacuum to reduce voids
2. Quantifying exposure of parts A and B
3. FTIR Measurement Procedures.
 - Direct measurement before cure
 - Direct measurement of fracture.
 - Use thinner surface samples to match bondline.
 - Apply chemometric analysis to analyze changes in spectra.



IGC Overview

‰ Technique to characterize physicochemical properties of materials

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